

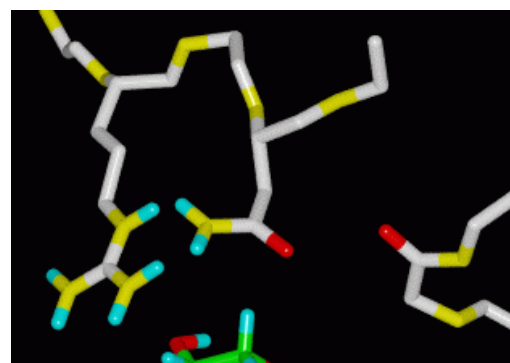
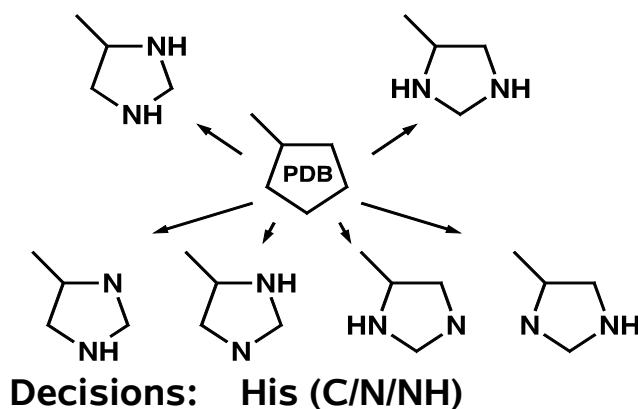
Interpreting Electron Densities: X-ray Structures

**“The X-ray Structure is an
Interpretation of the Electron Density
and thus Susceptible to
Plural Errors“**

Data from X-ray Measurements

- **Measured Data:** diffraction pattern
- **Derived Data:** electron density
- **Interpretation:** least squares fit of density via
 - spatial coordinates
 - temperature factors

- **Refined Data:** coordinates with assignment of atomic numbers (C, N, O, S etc.)



ASN (O/NH₂ wrong!)

⇒ **structure not obtained by measurement but interpretation**

Widely Accepted *Sloppy Usage* of Definitions

- Resolution: phys. limits due to crystal imperfections
NOT: precision of atomic positions
*e.g. bond angle deviation up to 10% in every structure
(angle $N-C_{\alpha}-C=O$ often 118°)*
- B-Factors: degree of freedom in interpretation
NOT: vibrational motion of atomic positions
*e.g. different B-factors for symmetry related atoms
(ortho-carbons in PHE or O/N in ASN)*
- R-Factor: measure of global regularity
NOT: quality of local coordinates
e.g. lowered by well positioned water molecules

⇒ structure is a fit to experimental and theoretical restraints

Statistical Quality Checks

Reference Geometry Parameters and Standard Deviations determined from Crystal Data (e.g. PROCHECK, WHAT_IF):

- Bond Lengths and Angles: Engh and Huber
- Dihedral Angles:
 - Ramachandran, ω
 - preferred conformations for χ_1, χ_2
- Close van der Waals Contacts
- Close Water Molecules

Actual Deviations of up to
! Four Times the Standard Deviations !
in the Statistical Analyses
are Tolerated

⇒ what means **4*S-DEV?** compare to force field references ...

Parameters from Crystal Data

Comparison: ENGH/HUBER-Parameters to AMBER-Force Field Reference Values

(Stat. Standard Deviation) and [Deviation of Ref. Values from Parameters]

	C-N	CA-C	CA-CB	N-CA	C-N-CA	CA-C-N	CB-CA-C	N-CA-C	N-CA-CB
Pro	1.341 (0.016)	-	-	1.466 (0.015)	122.60 (5.00)	116.90 (1.50)	-	111.80 (2.50)	103.00 (1.10)
Amber	1.335 [-0.006]	-	-	1.449 [-0.017]	121.90 [-0.70]	116.60 [-0.30]	-	110.10 [-1.70]	109.70 [+6.70]
Gly	-	1.516 (0.018)	-	1.451 (0.016)	120.60 (1.70)	116.40 (2.10)	-	112.50 (2.90)	-
Amber	-	1.522 [0.006]	-	1.449 [-0.002]	121.90 [1.30]	116.60 [0.20]	-	110.30 [-2.20]	-
Any	-	-	1.530 (0.020)	1.458 (0.019)	121.70 (1.80)	116.20 (2.00)	110.10 (1.90)	111.20 (2.80)	110.50 (1.70)
Amber	-	-	1.526 [-0.004]	1.449 [-0.009]	121.90 [0.20]	116.60 [0.40]	111.10 [1.00]	110.10 [-1.10]	109.50 [-1.00]

Lower AMBER-References: Strain Causes Higher Actual Value

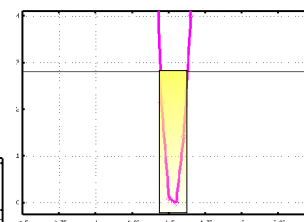
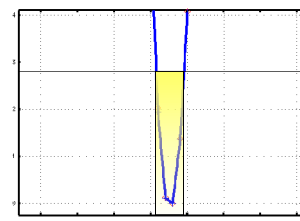
⇒ references agree! **4*S-DEV** means **0.08Å** (bonds), **8°** (angles)

Energy Cost for Deviation from References

Geometry Variations with 2.8 kcal/mol Energy Penalty

(e.g. $\frac{1}{2}$ twist-chair cyclohexane, Boltzmann factor 1:100)

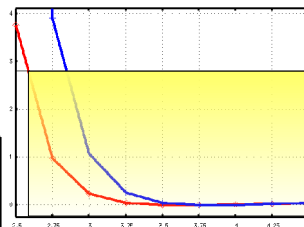
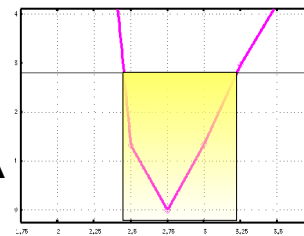
• **Single Bond:** $\pm 0.1 \text{ \AA}$
(C-C: 1.42 - 1.62 \AA)



• **Bond Angle:** $\pm 7.5^\circ$
(sp^3 Carbon: 102° - 117°)

• **Dihedral Angle:** **cis/trans peptide bond !**

• **vdW Repulsion:** -1.0 \AA
(C...C < 2.6 \AA)

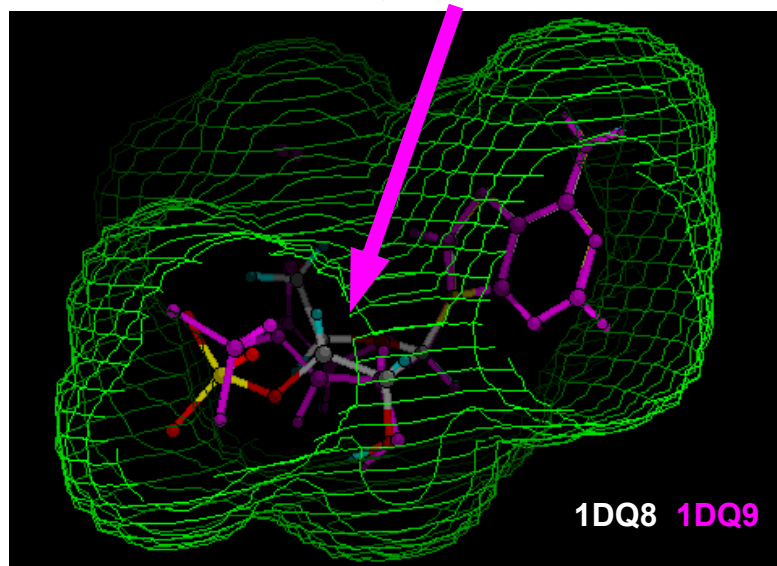


• **HBO Repulsion:** -0.3 \AA +0.5 \AA
(O...O: <2.5 \AA or > 3.3 \AA)

⇒ **NO geometry CAN have 4*S-DEV deviations; but they do!**

Other Discrepancies, Easier to Remedy?

- Conformation of Ambivalent Side Chains (e.g. ASN)
- Hetero-Groups
(e.g. Coenzyme A: superposition of structures from 1DQ8 and 1DQ9, in **1DQ9** with wrong stereochemistry!)



Does electron density fail to show the right stereoisomer?

⇒ **EVERY external information must be used, ALWAYS!**

Geometry: Minimum Requirements

- **Appropriate Parameters for Hetero-Groups**
- **Favourable Energy Contributions**
 - valence terms (bonds, angles)
 - van der Waals terms (no close contacts)
 - other non valence terms (no electronic repulsion)⇒ no energetical 'hot spots'
- **Optimum Hydrogen Bonding Network**
 - ambivalent side chain orientation
 - orientation of flexible hydrogens

**Requirements *Have to be Met*
Prior to Any Theoretical Study**

⇒ **CHEOPS** structure preparation takes care of this!